

INDUSTRIAL & POWER PLANT DESIGN ELEMENTS MODULE

1. In Refrigeration, how do you call a heat exchanger in which low-pressure refrigerant boils or vaporizes, thus absorbing heat that was removed from the refrigerated area by the cooling medium (water)?
 a) Evaporator b) Chiller c) Cooler d) Flooded Evaporator

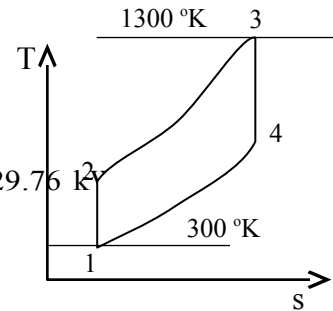
Answer: b) Chiller. In refrigeration application, the appropriate term used for evaporator is chiller.

2. In an ideal standard Brayton cycle, 1.5 kg/s of air at 101 kPa and 27 °C is compressed isentropically to a certain pressure and temperature after which the is added until the temperature becomes 1027 °C. Isentropic expansion occurs in the turbine. Determine the net power produced by the cycle.
 a) 629.56 kW b) 592.65 kW c) 529.76 kW d) 579.26 kW

Answer: c) 529.76 kW

Solution: $T_2 = T_4 = \sqrt{T_1 T_3} = \sqrt{(300)(1300)} = 624.5 \text{ °K}$

$$W_{\text{net}} = mc_p (T_3 - 2T_2 + T_1) = (1.5)(1.0062)[1300 - 2(624.5) + 300] = 529.76 \text{ kW}$$



3. In an air standard Otto cycle, the clearance volume is 12 % of the displacement volume. What is the thermal efficiency?
 a) 57 % b) 59 % c) 58 % d) 60 %

Answer: b) Thermal efficiency = 59 %

Solution:

$$V_2 = 0.12 V_D \quad r_k = \frac{V_1}{V_2} = \frac{V_2 + V_D}{V_2} = \frac{0.12 V_D + V_D}{0.12 V_D} = \frac{1.12}{0.12} = 9.33$$

$$e_{\text{th}} = \left[1 - \frac{1}{(r_k)^{k-1}} \right] (100 \%) = \left[1 - \frac{1}{(9.33)^{0.4}} \right] (100 \%) = 59.08 \%$$

4. In an air-standard Brayton cycle, air enters compressor at 1 bar and 15 °C. The pressure leaving the compressor is 0.6 MPaa and maximum temperature of the cycle is 1000 °C. What is the maximum net work, in kJ/kg?
 a) 319.52 b) 392.51 c) 315.29 d) 352.19

Answer: d) Max $W_{\text{net}} = 352.19 \text{ kJ/kg}$

Solution: $T_2 = \sqrt{T_1 T_3} = \sqrt{(15 + 273)(1000 + 273)} = 605.49 \text{ °K}$

$$\text{Max } w_{\text{net}} = c_p (T_3 - 2T_2 + T_1) = (1.0062)[1273 - 2(605.49) + 288] = 352.19 \text{ kJ / kg}$$

5. What is the clearance volumetric efficiency of an ammonia compressor designed with 4 % clearance and operating between condenser temperature of 30 °C ($p_{\text{sat}} = 1.1672 \text{ MPaa}$) and evaporator temperature of 4 °C ($p_{\text{sat}} = 497.48 \text{ kPa}$)?
 a) 93.61 % b) 93.68 % c) 96.31 % d) 96.83 %

Answer: c) Volumetric Efficiency = 96.31 %

Solution: $\eta_v = \left[1 + c - c \left(\frac{p_d}{p_s} \right)^{\frac{1}{k}} \right] (100 \%) = \left[1.04 - 0.04 \left(\frac{1167.2}{497.48} \right)^{\frac{1}{1.304}} \right] (100 \%) = 96.31 \%$

Note: $k = 1.304$, for ammonia (R-717) $k = 1.30$, for R-134a

6. The percent rating of water tube boiler is 200 %, factor of evaporation is 1.10, and heating surface is 400 ft². Determine the rate of evaporation, in kg/hr.

a) 1831 b) 1831 c) 1138 d) 1813

Answer: c) Rate of evaporation = 1138 kg/hr

$$\text{Solution: Rated Bo. Hp} = \frac{\text{H.S.}}{10} = \frac{400}{10} = 40$$

$$\text{Dev. Bo. Hp} = \text{Percent Rating} (\text{Rated Bo. Hp}) = 2 (40) = 80$$

$$\text{Dev. Bo. Hp} = m_s (h_1 - h_B) = m_s (2257) \text{ F.E.}$$

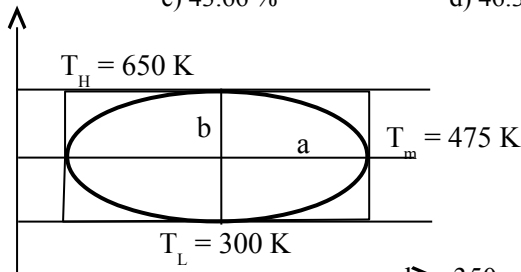
$$m_s = \frac{\text{Dev. Bo. Hp}}{2257 \text{ FE}} = \frac{80(35314)}{2257(1.1)} = 1138 \text{ kg / hr}$$

7. A Carnot cycle is represented by a rectangle in a Ts diagram that operates between temperature limits of 300 °K and 650 °K. Inscribed within a rectangle is a ellipse of maximum major and minor axes, represents a cycle and operating at the same temperature limits. Considering that the major axis of the ellipse is two times that of its minor axis. Determine the thermal efficiency of the cycle represented by an ellipse.

a) **44.88 %** b) 48.48 % c) 43.66 % d) 46.36 %

Answer: a) 44.88 %

Solution:



$$a = 2b = 650 - 300 = 350 \text{ }^\circ\text{K} \qquad b = \frac{350}{2} = 175 \text{ }^\circ\text{K}$$

$$A = \pi ab = \pi (350) (175) = 61\,250\pi = W_{\text{net}} \qquad \Delta S = 2a = 700$$

$$Q_A = T_m (\Delta S) + \frac{W_{\text{net}}}{2} = 475 (700) + \frac{61\,250\pi}{2} = 428\,711.275 \text{ kJ}$$

$$e = \frac{W_{\text{net}}}{Q_A} \times 100\% = \frac{61\,250\pi}{428\,711.275} (100\%) = 44.88 \%$$

8. Determine the critical radius in cm for an asbestos-cement covered pipe [$k_{\text{asb}} = 0.208 \text{ W/m}\cdot^\circ\text{K}$]. The external heat-transfer coefficient is 1.5 Btu/h-ft²-°F.

a. 2.44 cm b. 2.55 cm c. 2.66 cm d. 2.22 cm

Solution: a) Critical radius = 2.44 cm

$$\text{Solution: } r_c = \frac{k}{h_c} = \frac{0.208 \frac{\text{W}}{\text{m}\cdot^\circ\text{K}}}{\left(1.5 \frac{\text{Btu}}{\text{hr}\cdot\text{ft}^2\cdot^\circ\text{F}}\right)(5.675)} = 0.0244 \text{ m} = 2.44 \text{ cm}$$

$$\text{Note: } 1 \frac{\text{Btu}}{\text{hr}\cdot\text{ft}^2\cdot^\circ\text{F}} = 5.675 \frac{\text{W}}{\text{m}^2\cdot^\circ\text{F}}$$

9. A hiker carried an Aneroid barometer from the foot of Mount Banahaw to the camp of the NPA leader Ka Roger. On the foot of the mountain, the barometer reads 30.150 inches of Hg, and on the camp, which is nearly at the top of the mountain it reads

28.607 inches of Hg. Assume that the average air density (atmospheric) was 0.075 pcf, estimate the height of the mountain, in ft.

- a) 1455.54 b) 1545.54 c) 1554.54 d) 1455.55

Answer: a) 1455.54 ft

Solution:

$$h = \frac{P_1 - P_2}{\rho g} = \frac{(30.150 - 28.607 \text{'' Hg}) \left(\frac{14.7 \text{ psi}}{29.92 \text{'' Hg}} \right) (144)}{0.075} = 1455.54$$

10. In a geothermal power plant, the mass flow rate of ground water is 4000 kg/s and the quality after throttling is 20%. If the turbine power is 80 MW, what is the change in enthalpy of steam at the inlet and outlet of the turbine?

- a) 120 kJ/kg b) 100 kJ/kg c) 200 kJ/kg d) 150 kJ/kg

Answer: b) Change of enthalpy = 100 kJ/kg

Solution: $m_s = x_3 m_c = (0.20)(4000) = 800 \text{ kg/s}$ $\Delta h = \frac{W_t}{m_s} = \frac{80000}{800} = 100 \text{ kJ/kg}$

11. Determine the partial pressure of CO₂ for the following analysis of gas mixture by weight at 101 kPaa and 25 °C: CO₂ = 35 % and N₂ = 65 %.

- a) 25.75 kPaa b) 27.55 kPaa c) 52.75 kPaa d) 52.57 kPaa

Answer: a) partial pressure of CO₂ = 25.75 kPaa

Solution: For 100 kg of the mixture, total number of moles,

$$n_T = \frac{CO_2}{44} + \frac{N_2}{28} = \frac{35}{44} + \frac{65}{28} = 0.7955 + 2.3214 = 3.12 \text{ kgmoles}$$

From the equation, $p v = n \bar{R} T$ $\frac{\bar{R} T}{v} = \frac{p_{CO_2}}{n_{CO_2}} = \frac{P_T}{n_T}$

$$p_{CO_2} = \left(\frac{n_{CO_2}}{n_T} \right) P_T = \left(\frac{0.7955}{3.12} \right) (101) = 25.75 \text{ kPaa}$$

12. A certain coal has the following ultimate analysis by weight. C = 67 %, Ash = 5 %, Moisture = 8 %, N = 6 %, H = 3 %, Sulfur = 7 %, O = 4 %. Calculate the higher heating value in Btu/lb.

- a) 15 179 b) 11 579 c) 17 519 d) 19 517

Answer: b) 11 579 Btu/lb

Solution:

$$HHV = 14544C + 62028 \left(H_2 - \frac{O_2}{8} \right) + 4050S = (14544)(0.67) + (62028) \left(0.03 - \frac{0.04}{8} \right) + (4050)(0.07)$$

$$HHV = 11578.68 \text{ Btu/lb}$$

13. Determine the heating value of a certain fuel with a SG = 0.997.

- a) 42 975 kJ/kg b) 42 597 kJ/kg c) 42 795 kJ/kg d) 42 579 kJ/kg

Answer: a) 42 975 kJ/kg

Solution: $HHV = 51716 - 8793.8(SG)^2 = 51716 - 8793.8(0.997)^2 = 42975 \text{ kJ/kg}$

14. If an airplane is flying at an altitude of 5 800 ft, what is the approximate atmospheric pressure at the said altitude?

- a) 12.34 psi b) 14.7 psi c) 13.24 psi d) 14.32 psi

Answer: a) 12.34 psi

Solution: $p = 29.92 \text{ in Hg} - \left(\frac{1 \text{'' Hg}}{1000 \text{ ft}} \right) (4800 \text{ ft}) = 25.12 \text{ in Hg} = 12.34 \text{ psia}$

Answer: b) Evaporative condenser

21. A boiler operates at 82 % efficiency while the mass of steam generated is 490 200 kg in 6 hours. The enthalpy of steam is 3187 kJ/kg and feed is 604.83 kJ/kg while the fuel used for boiler has a heating value of 32 567.85 kJ/kg. Find the mass of fuel needed per day in metric tons.
a) 179.6 b) 189.6 c) 198.6 d) 169.8

Answer: b) Fuel consumption per day = 189.6 metric tons

$$\text{Solution: } m_s = \frac{490\,200}{6} = 81\,700 \text{ kg / hr}$$

$$Q_F = \frac{m_s(h_2 - h_1)}{E_B} = \frac{81\,700(3187 - 604.83)}{0.82} = 257\,272\,303.7 \text{ kg / hr}$$

$$m_F = \frac{Q_F}{q_h} = \frac{257\,272\,303.7}{32\,567.85} = 7\,899.58 \text{ kg / hr} = 189\,589.8958 \text{ kg / day} = 189.6 \text{ Metric tons / day}$$

22. It refers to a factor used in calculating the over-all heat transfer through the tube walls of a condenser tube or other heating surface. It includes the sum of the heat-transfer rate of the layer of dirt and foreign material that builds up on the water side of the tube. What is this factor?
a) Cooling factor b) Contact factor c) By-pass factor d) Fouling factor

Answer: d) Fouling Factor

23. A 10 kg/s of air enters the theater at 16 °C. The theater is to be maintained at 27 °C DB and 20 °C WB. If the sensible heat ratio is 0.71, what is the latent heat load of the theater?
a) 45.21 kW b) 54.21 kW c) 110.682 kW d) 42.51 kW
Answer: a) latent heat load, $q_L = 45.21 \text{ kW}$

$$\text{Solution: } q_s = 1.0062m(t_r - t_s) = (1.0062)(10)(27 - 16) = 110.682 \text{ kW}$$

$$q_L = \frac{q_s}{\text{SHR}} - q_s = \frac{110.682}{0.71} - 110.682 = 45.21 \text{ kW}$$

24. A 80 MW power plant has an average load of 34 500 kW and a load factor of 0.75. Find the reserve power over a peak load power.
a) 14 000 kW b) 34 000 kW c) 24 000 kW d) 4 000 kW

Answer: b) Reserve Power = 34 000 kW

$$\text{Solution: } \text{Load Factor} = \frac{\text{Average Load}}{\text{Peak Load}} \quad \text{Peak Load} = \frac{\text{Average Load}}{\text{Load Factor}} = \frac{34\,500}{0.75} = 46\,000 \text{ kW}$$

$$\text{Reserve Power} = \text{Plant Capacity} - \text{Peak Load} = 80\,000 - 46\,000 = 34\,000 \text{ kW}$$

25. What is the standardized term used by the industry to describe any device that meters or regulates the flow of liquid refrigerant to an evaporator?
a) Refrigerant control b) Expansion valve c) Throttling valve d) Capillary tube

Answer: b) Expansion Valve

26. A steam generator burns fuel oil with 25 % excess air. Fuel may be represented by $C_{14}H_{30}$. Calculate the actual-air fuel ratio.
a) 17.63 b) 18.63 c) 16.63 d) 15.63

Answer: b) Actual A/F ratio = 18.63 kg air per kg fuel

$$\text{Solution: } A : F = \frac{137.28(n + 0.25m)(1 + e)}{12n + m} = \frac{137.28[14 + 0.25(30)]}{12(14) + 30} (1.25) = 18.63$$

Note: Equation above is used for hydro-carbon fuels, C_nH_m .

27. A container filled with helium is dropped 3000 m above the ground, find the change of temperature?

- a) 12.43 °C b) 9.43 °C c) 15.43 °C d) 8.43 °C

Answer: b) The change of temperature, $\Delta T = 9.43$ °C

$$\text{Solution: } \Delta P = \Delta KE = \Delta U \quad \frac{mg(\Delta z)}{1000} = mc_v(\Delta T) = m \left(\frac{R}{k-1} \right) (\Delta T) \quad \Delta T = \frac{(k-1)g(\Delta z)}{1000R} \quad k = 1.666$$

$$R = \frac{\bar{R}}{M} = \frac{8.3143}{4} = 2.078575 \text{ kJ/kg} \cdot ^\circ \text{K} \quad \Delta T = \frac{0.666(9.8066)(3000)}{1000(2.078575)} = 9.43 \text{ } ^\circ \text{C}$$

Note: For Helium, $k = 1.666$ and $M = 4$ kg/kgmole

28. A 3153-lb car is accelerated from 32 fps to 55 fps in 10 seconds. Determine the work done, in Btu.

- a) 125.92 b) 192.52 c) 152.92 d) 129.52

Answer: a) Work done, $W = 125.92$ Btu

$$\text{Solution: } W = \Delta K = \left(\frac{m}{2} \right) (V_2^2 - V_1^2) = \frac{3153[(55)^2 - (32)^2]}{2(32.2)(778)} = 125.92 \text{ Btu}$$

Note: 1 Btu = 778 ft-lb & 1 lb_f = 32.2 lb_m-fps²
Work = Kinetic energy

29. A turbine has a peripheral coefficient of 0.6. Find the runner diameter of the turbine if it operates at 450 rpm and a head of 60 m.

- a) 0.874 m b) 0.784 m c) 0.478 m d) 0.748 m

Answer: a) Water runner diameter, $D_r = 0.874$ m

$$\text{Solution: Peripheral Coefficient, } \Phi = \frac{\text{Peripheral velocity}}{\text{Velocity of jet}} = \frac{\pi D_r N}{60\sqrt{2gH_{\text{eff}}}}$$

$$D_r = \frac{\Phi(60\sqrt{2gH_{\text{eff}}})}{\pi N} = \frac{(0.60)(60)\sqrt{2(9.8066)(60)}}{\pi(450)} = 0.874 \text{ m}$$

30. Heat is transferred from hot water to an oil in a double-pipe counter-flow heat exchanger. Water enters the outer pipe at 120 °C and exits at 55 °C while the oil enters the inner pipe at 26 °C and exits at 65 °C. What is the log-mean temperature difference (LMTD)?

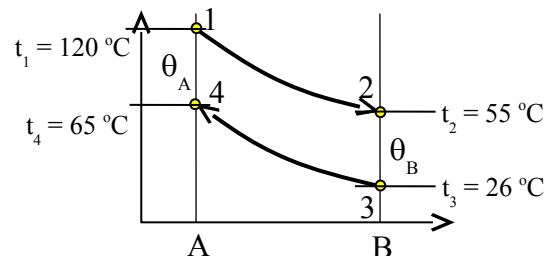
- a) 42.60 °C b) 40.62 °C c) 46.20 °C d) 42.06 °C

Answer: b) Log Mean Temperature Difference, LMTD = 40.62 °C

Solution:

$$\theta_A = t_1 - t_4 = 120 - 65 = 55 \text{ } ^\circ \text{C} = \theta_{\text{max}}$$

$$\theta_B = t_2 - t_3 = 55 - 26 = 29 \text{ } ^\circ \text{C} = \theta_{\text{min}}$$



$$LMTD = \frac{\theta_{\max} - \theta_{\min}}{\ln\left(\frac{\theta_{\max}}{\theta_{\min}}\right)} = \frac{55 - 29}{\ln\left(\frac{55}{29}\right)} = 40.62 \text{ }^{\circ}\text{C}$$

31. In a Rankine cycle, steam enters the turbine at 2.5MPa and a condenser pressure of 50KPa. What is the quality of steam at the turbine exhaust? Steam Properties: @ 2.5MPaa, $h = 2803.1 \text{ kJ/kg}$ & $s = 6.2575 \text{ kJ/kg-}^{\circ}\text{K}$; @ 50kPaa, $h_f = 340.49 \text{ kJ/kg}$, $h_{fg} = 2305.4 \text{ kJ/kg}$, $v_f = 0.00103 \text{ m}^3/\text{kg}$, $s_f = 1.0910 \text{ kJ/kg-}^{\circ}\text{K}$, $s_{fg} = 6.5029 \text{ kJ/kg-}^{\circ}\text{K}$.
- a) 79.45 % b) 97.45 % c) 59.75 % d) 95.55 %

Answer: a) Steam Quality, $x_2 = 79.45 \%$

$$\text{Solution: } x_2 = \left(\frac{s_2 - s_{f2}}{s_{fg}} \right) 100\% = \left(\frac{6.2575 - 1.0910}{6.5029} \right) (100\%) = 79.45 \%$$

32. An adiabatic turbine steam generating plant receives steam at a pressure of 7.0 MPa and 550°C ($h = 3531 \text{ kJ/kg}$) and exhausts at a condenser pressure of 20kPa ($h = 2290 \text{ kJ/kg}$). The turbine inlet is 3 meters higher than the turbine exit, inlet steam velocity is 15m/s and the exit is 300m/s. Calculate the turbine work in kJ/kg.
- a) 1296.14 b) 1196.24 c) 1619.42 d) 1294.16

Answer: b) The turbine work, $W_t = 1196.24 \text{ kJ/kg}$

$$\text{Solution: } W_t = -(\Delta h + \Delta k + \Delta P) = (h_1 - h_2) + \frac{(V_1^2 - V_2^2)}{2000} + \frac{g(z_2 - z_1)}{1000}$$

$$W_t = (3531 - 2290) + \frac{(15)^2 - (300)^2}{2000} + \frac{9.8066(3)}{1000} = 1196.24 \text{ kJ/kg}$$

33. Determine the indicated power of a four-cylinder, 4-stroke, Diesel engine with 20-cm bore and 30-cm stroke running at 1000 rpm and has a reading of 450 kPa mean effective pressure in the indicator diagram.
- a) 159.83 Hp b) 189.53 Hp c) 158.93 Hp d) 198.53 Hp

Answer: b) Indicated power, $IP = 189.53 \text{ Hp}$

Solution: Volume displacement,

$$V_D = \frac{\pi}{4} D^2 L N n_c = \frac{\pi}{4} D^2 L \left(\frac{n}{2} \right) n_c / 60 = \left(\frac{\pi}{4} \right) (0.20)^2 (0.30) \left(\frac{1000}{2} \right) \left(\frac{4}{60} \right) = 0.3142 \text{ m}^3 / \text{s}$$

$$\text{Indicated Power, } W_I = p_I V_D = (450)(0.3142) = 141.4 \text{ kW} = 189.53 \text{ Hp}$$

34. The approach and efficiency of cooling tower are 10°C and 65 %, respectively. If the temperature of water leaving the tower is 27°C , determine the temperature of water entering the tower.
- a) 54.57°C b) 55.47°C c) 45.57°C d) 54.75°C

Answer: c) Temperature of water entering the tower, $t_3 = 45.57^{\circ}\text{C}$

$$\text{Solution: } t_{wbl} = t_4 - CA = 27 - 10 = 17^{\circ}\text{C}$$

$$\text{Tower Efficiency, } e_t = \frac{ACR}{TCR} = \frac{t_3 - t_4}{t_3 - t_{wbl}}$$

$$t_3 = \frac{t_4 - e t_{wbl}}{1 - e} = \frac{27 - 0.65(17)}{1 - 0.65} = 45.57^{\circ}\text{C}$$

35. An air-vapor mixture has a dry bulb temperature of 30°C and a humidity ratio of 0.015 kg/kg d.a., calculate the enthalpy of the moist air.

- a) 68.527 kJ/kg d.a b) 86.527 kJ/kg d.a c) 65.827 kJ/kg d.a. d) 67.528 kJ/kg d.a

Answer: a) Enthalpy, $h = 68.527 \text{ kJ/kg d.a.}$

Solution: Using the IHVE formula, $h = 1.007t - 0.026 + W(2501 + 1.84t)$

$$h = 1.007(30) - 0.026 + (0.015)[2501 + 1.84(30)] = 68.527 \text{ kJ / kg da}$$

36. Determine the equivalence ratio of the following mixture: 4 grams of butane, C_4H_{10} , and 75 grams of air.

- a) 0.72 b) 0.92 c) 0.62 d) 0.82

Answer: d) Equivalence ratio = 0.82

Solution: Theoretical air-fuel ratio, $W_a = \frac{137.28(n + 0.25m)}{12n + m} = \frac{137.28[4 + 0.25(10)]}{12(4) + 10} = 15.385 \text{ kg / kg}$

Actual Air-fuel ratio, $W_{aa} = \frac{75}{4} = 18.75 \text{ kg / kg}$

$$\text{Equivalence ratio} = \frac{W_a}{W_{aa}} = \frac{1}{1 + e} = \frac{15.385}{18.75} = 0.82$$

37. Find the air H_p of an industrial fan that delivers $25 \text{ m}^3/\text{s}$ of air through a 900 mm by 1200 mm outlet. Static pressure is 127 mm of water gage and air density is 1.18 kg/m^3 .

- a) 52.3 Hp b) 35.2 Hp c) 42.3 Hp d) 34.2 Hp

Answer: a) Fan power, $W_F = 52.3 \text{ Hp}$

Solution: $V = \frac{Q}{A} = \frac{25}{(0.9)(1.2)} = 23.15 \text{ m / s}$ $h_s = \frac{\rho_w}{\rho_a} h_w = \left(\frac{1000}{1.18}\right)(0.127) = 107.63 \text{ m air}$

$$h_v = \frac{V^2}{2g} = \frac{(23.15)^2}{2(9.8066)} = 27.32 \text{ m air} \quad h_t = h_s + h_v = 107.63 + 27.32 = 134.95 \text{ m air}$$

$$W_F = \frac{\rho g Q h_t}{1000} = \frac{1.18(9.8066)(25)(134.95)}{1000} = 39.04 \text{ kW} = 52.3 \text{ Hp}$$

38. Determine the indicated mean effective pressure of an engine, in psi, having a brake mean effective pressure of 750 kPa and 80 % mechanical efficiency.

- a) 138 b) 137 c) 136 d) 135

Answer: c) Indicated mean effective pressure, $p_i = 136.01 \text{ psi}$

Solution: $\eta_m = \frac{W_B}{W_I} (100 \%) = \frac{p_B}{p_I} (100 \%) \quad p_I = \frac{p_B}{\eta_m} = \frac{750}{0.80} = 937.5 \text{ kPa} = 136.01 \text{ psi}$

39. The indicator card (actual p-V diagram) of an engine in a Diesel power plant indicates an area of 0.06 m^2 and length of 300 mm, and with a spring scale of 2500 kPa/m. The engine was tested using Prony brake with lever arm of 3 m and tare weight of 8 kN. Determine the mechanical efficiency if the engine is running at 600 rpm. The engine is 2-stroke and has 12 cylinders, 300 mm bore, and 450 mm stroke.

- a) 78.01 % b) 79.01 % c) 82.01 % d) 76.01 %

Answer: b) Engine mechanical Efficiency = 79.01 %

Solution: $W_B = \frac{\pi n T}{30} = \frac{\pi(600)(8 \text{ kN})(3 \text{ m})}{30} = 1507.96 \text{ kW}$ $h = \frac{A}{b} = \frac{0.06 \text{ m}^2}{0.3 \text{ m}} = 0.2 \text{ m}$

$$p_i = k_s h = (2500 \text{ kPa / m})(0.2 \text{ m}) = 500 \text{ kPa}$$

Answer: b) RH = 55 % to 60 %

45. When adding the refrigerant in a refrigeration system, it shall be charge into which of the following parts of the system?
- a) High pressure side
 - b) Low pressure side
 - c) Low and high pressure side
 - d) Compressor discharge line

Answer: b) Low pressure side

46. A 1.5 MW Diesel power generating unit has a generator efficiency of 85 %. Determine the volume flow rate, in lps, of cooling water required for the engine at 18 °C temperature rise.
- a) 21 lps
 - b) 19 lps
 - c) 22 lps
 - d) 23 lps

Answer: a) Volume flow = 21 lps

Solution:
$$BP = \frac{W_k}{\eta_g} = \frac{1500}{0.85} = 1764.71 \text{ kW}$$

Heat Balance for a Typical Diesel Engine at Full Load

Brake Power	34 %
Cooling System Losses	30 %
Exhaust Losses	26 %
Friction, Radiation, etc	10 %

$$Q_{cw} = \frac{BP}{0.34}(0.30) = \left[\frac{1500}{0.85(0.34)} \right](0.30) = 1557.1 \text{ kW}$$

$$m_{cw} = \frac{Q_{cw}}{c_p(\Delta T)} = \frac{1557.1}{4.187(18)} = 20.66 \text{ kg / s}$$

$$V_{cw} = \frac{20.66 \text{ kg / s}}{1 \text{ kg / li}} = 20.66 \text{ lps}$$

47. Why is a boiler feed tank heated to approximately 85 °C?
- a) To reduce the energy required to raise steam
 - b) To reduce the content of total dissolved solids in the water supplied to the boiler
 - c) To reduce the gas content of the water
 - d) To reduce the content of suspended solids in the water

Answer: c) To reduce the gas content of the water

48. What is used to dry air?
- a) A separator
 - b) A strainer
 - c) A steam trap
 - d) A tee piece

Answer: a) A separator

49. A Carnot refrigerator is to remove heat from a cooled space at a rate of 18 000 kJ/hr to maintain the temperature at – 8 °C. If the air surrounding the refrigerator is 25 °C, determine the minimum power required for the refrigerator.
- a) 0.723 kW
 - b) 0.523 kW
 - c) 0.623 kW
 - d) 0.423 kW

Answer: c) 0.623 kW

Solution:
$$W = \frac{Q_L(T_H - T_L)}{T_L} = \frac{18000(25 + 8)}{(273 - 8)} = 2241.51 \text{ kJ / hr} = 0.623 \text{ kW}$$

50. A Pelton wheel has a capacity of 25 000 kW with head of 290 m. What is the number of jet needed for this turbine if the jet if the nozzle jet diameter is 200 mm.
- a) 3
 - b) 4
 - c) 2
 - d) 1

Answer: b) Number of Jets or Nozzles = 4

$$\text{Solution: } Q = \frac{1000 W_t}{\rho g H_T} = \frac{(25\,000)(1000)}{(1000)(9.8066)(290)} = 8.79 \text{ m}^3 / \text{s}$$

$$A = \left(\frac{\pi}{4}\right) D^2 = \left(\frac{\pi}{4}\right) (0.20)^2 = 0.03142 \text{ m}^2$$

$$V = \sqrt{2gH} = \sqrt{2(9.8066)(290)} = 75.42 \text{ m/s} \quad A_T = \frac{Q}{V} = \frac{8.79}{75.42} = 0.1165 \text{ m}^2$$

$$\text{No. of jets: } N_j = \frac{0.1165 \text{ m}^2}{0.03142 \text{ m}^2 / \text{jet}} = 3.71 \text{ jets Use 4 jets}$$