

FLUID MECHANICS

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***FLUID MECHANICS**

*It is a physical science dealing with the action at rest or in motion and with application and devices in engineering using fluids

Division of Fluid Mechanics

1. FLUID STATICS - deals with fluid at rest

2. FLUID DYNAMICS

- concerned with fluids in motion

***TYPES OF FLUIDS**

1. Ideal Fluids

2. Real Fluids

1. Ideal Fluids

- *Assumed to have no viscosity**
- *Incompressible**
- *Have uniform velocity when flowing**
- *No eddy currents or turbulence**

2. Real Fluids

- *Exhibit finite viscosities**
- *Non-uniform velocity distribution flowing**
- *Compressible**
- *Experience friction and turbulence**

FLUIDS

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graph TD; FLUIDS[FLUIDS] --> Ideal[Ideal Fluids]; FLUIDS --> Real[Real Fluids]; Real --> Newton[Newton Fluids]; Real --> NonNewtonian[Non-Newtonian Fluids]; NonNewtonian --> Pseudo[Pseudo Plastic Fluids]; NonNewtonian --> Delatual[Delatual Fluids]; NonNewtonian --> Bingham[Bingham Fluids];
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Ideal Fluids

Real Fluids

Newton Fluids

Non-Newtonian
Fluids

Pseudo Plastic
Fluids

Delatual Fluids

Bingham Fluids

*FLUID PROPERTIES

1. Mass Density
2. Specific Volume
3. Specific weight
4. Specific Gravity
5. Absolute Viscosity or Viscosity
6. Kinematic Viscosity
7. Bulk Modulus
8. Compressibility
9. Coefficient Thermal Expansion

* MASS DENSITY (ρ , rho)

* The mass per unit volume
- The mass per unit volume

$$\rho \equiv m/v$$

Where: ρ , mass density units; BG slugs/ft³
Where: ρ , mass density units; SI kg/m³
m, mass SI kg/m³
v, volume SI kg/m³

For ideal gas

$$\rho \equiv \frac{P}{RT}$$

where: P, absolute pressure of gas in Pascal
R, gas constant (Air= 287J/kg·K)
T, temperature in Kelvin or Rankine units

*Specific Volume (Vs)

- *It is defined as volume per unit mass.
- It is defined as volume per unit mass.
- It is the reciprocal of density.
- It is the reciprocal of density.

$$V_s = \frac{v}{m} = \frac{1}{\rho}$$

*Specific Weight

*It is the weight per unit volume
-It is the weight per unit volume

$$\gamma = \frac{w}{vol} = \frac{mg}{vol}$$

Units: BG lb/ft³

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SI N/m³

SI N/m³

* Specific Gravity (SG)

- * It is defined as the ratio of the density of the fluid to the density of water at some specified temperature. Normally the standard temperature is taken at 4°C (392°F)

$$SG = \frac{\rho_F}{\rho_w} = \frac{\gamma_F}{\gamma_w}$$

*Absolute Viscosity or Viscosity, μ (μ)

$$\mu = \frac{T}{dV/dy}$$

Where: T, shear stress

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μ , absolute viscosity

μ , absolute viscosity

y, distance between plates

y, distance between plates

V, velocity

V, velocity

* Kinematic Viscosity, ν

(ν)

* $\nu = \mu / \rho$ Where: μ , absolute viscosity
 ρ , density

System	Absolute, μ	Kinematic, ν
British	lb·sec/ft ² slug/ft·sec	ft ² /sec
Metric	dyne·s/cm ² (Poise)	cm ² /s (Stoke)
S.I.	Pa·Sec (N·s/m ²)	m ² /s

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*Note:

$$1 \text{ Poise} = 1 \text{ dyne}\cdot\text{sec}/\text{cm}^2 = 0.1 \text{ Pa}\cdot\text{s}$$

$$1 \text{ Stoke} = 0.0001 \text{ m}^2/\text{s}$$

$$1 \text{ dyne} = 10^{-5} \text{ N}$$

*Bulk Modulus

*It is a measure of fluid compressibility.

-It is the ratio of the change in pressure in the corresponding change in volume

$$E_v = \frac{-dP}{dVol/Vol}$$

where; dP, the change in pressure needed to create a change in volume

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dVol, the resulting change in volume

Vol, initial volume

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$$E_v = \frac{-dP}{d\rho/\rho}$$

Where: dP, the change in pressure

dρ, the change in fluid density

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dρ, the change in fluid density

ρ, initial density

*Compressibility (β)

* Also known as the coefficient of compressibility

- It is the fractional change in volume of fluid per unit change in pressure.
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$$- \beta = \frac{1}{Ev} = \frac{-\frac{dVol}{Vol}}{dP} = \frac{-d\rho/\rho}{dP} = \frac{-dY/Y}{dP}$$

* Coefficient of Thermal Expansion, α_T

* It represents the ability of fluid to expand or compress under the influence of increase in temperature

$$\underline{\underline{\alpha_T}} = \frac{dVol/Vol}{dT} = \frac{1}{vol} \cdot \frac{dVol}{dT}$$

$$\alpha_T = \frac{d\rho/\rho}{dT}$$

SURFACE TENSION

Liquids have cohesion and adhesion which are forms of molecular attraction. Cohesion permits a liquid to resist tensile stress and adhesion permits it to adhere to another body. The surface tension, σ is defined as the intensity of the molecular attraction per unit length along any line in the surface. It has the dimensions of force per unit length. Its units are **dyne/cm** or **N/m**.

CAPPILLARITY

Caused by both cohesion and adhesion. When the effect of cohesion is less than adhesion, the liquid will wet a solid surface which is in contact and rise to the point of contact. If the effect of cohesion is greater than that of adhesion the liquid surface will be depressed.

For clean tube, $\theta = 0^\circ$

For Hg, $\theta = 140^\circ$

For tubes greater than 12mm, the cappillary effects are neglected.

VAPOR PRESSURE

Molecular activity in liquid will allow some of the molecules to escape the liquid surface. Molecules of the vapor also condense back into liquid. The vaporization and condensation at contact temperature as equilibrium processes. The equilibrium exerted by these free molecules is known as vapor pressure or saturation pressure.

PROBLEMS

* A reservoir of glycerin has a mass of 1200 kg and a volume of 0.952 m³. Find its (a) weight, (b) unit weight, (c) mass density and (d) specific gravity.

Given: $m = 1200 \text{ kg}$

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 $\text{Vol} = 0.952 \text{ m}^3$

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$$(a) W = mg = 1200 (9.81)$$

$$(a) W = mg = 11,772 \text{ N}$$

$$(b) \gamma = \frac{W}{\text{vol}} = \frac{11,772 \text{ N}}{0.952 \text{ m}^3}$$

$$= 12,365.55 \text{ N/m}^3$$

$$(b) = 0.952 \text{ m}^3$$

$$(c) \rho = \frac{m}{\text{vol}} = \frac{1200 \text{ kg}}{0.952 \text{ m}^3}$$

$$(d) SG = \frac{\rho_f}{\rho_w} = \frac{1260.5 \text{ kg/m}^3}{1000}$$

$$= 1.26$$

$$(d) SG = \frac{1260.5}{1000}$$

$$= 1.26$$

* 2: If the specific volume of certain gas is $0.7848 \text{ m}^3/\text{kg}$, what is its specific weight?

$$\rho = \frac{1}{v_s} = \frac{1}{0.7848} = 1.27 \text{ kg/m}^3$$

$$\begin{aligned} \gamma &= \rho g = 1.27 (9.81) \\ &= 12.46 \text{ N/m}^3 \\ &= 12.46 \text{ N/m}^3 \end{aligned}$$

* 3. Find the mass density of helium at 44°C with a pressure of 184 kPa absolute. $R = 2079 \text{ J/kg} \cdot \text{K}$

$$\rho \equiv \frac{P}{RT} = \frac{184 \times 10^3}{2079(44+273)} = 0.5 \text{ kg/m}^3$$

* 4. Air is kept at a pressure of 200 kPa absolute and a temperature of 30°C in a 500 L container. What is the mass of air?

$$\rho = \frac{P}{RT} = \frac{200 \times 10^3}{287(30+273)} = 2.3 \text{ kg/m}^3$$

$$\rho = \frac{m}{v}; m = \rho v = 2.3(500 \times 10^{-3}) = 1.15 \text{ kg}$$

* 5. A cylindrical tank of 80 cm diameter and 90 cm high is filled with the liquid. The weight of liquid is 420 kg. The weight of the tank is 40 kg. What is the unit weight of liquid in KN/m³?

$$\text{Mass of liquid} = 420 - 40 = 380 \text{ kg}$$

$$\text{Vol} = \pi D^2 h / 4 = 0.45 \text{ m}^3$$

$$\begin{aligned} \gamma &= \frac{mg}{\text{vol}} = 380(9.81) / 0.45 \\ &= 8.284 \text{ KN/m}^3 \end{aligned}$$

* If the viscosity of water at 70°C is 0.00402 poise and its specific gravity is 0.978. Determine its absolute viscosity in Pa·s and its kinematic viscosity in m²/s and stoke.

$$1 \text{ Stoke} = 0.001 \text{ m}^2/\text{s}$$

Note: 1 Poise = 0.1 Pa·s

$$\mu = 0.00402 \text{ Poise} \times 0.1 \text{ Pa}\cdot\text{s}/\text{Poise}$$

$$\mu = 0.000402 \text{ Pa}\cdot\text{s}$$

Absolute Viscosity,

$$V = \frac{0.000402 \text{ Poise}}{0.978} \times 0.1 \text{ Pa}\cdot\text{s}/\text{Poise}$$

$$V = 4.11 \times 10^{-7} \text{ m}^2/\text{s}$$

Kinematic Viscosity

=

$$V = 4.11 \times 10^{-7} / \text{s}$$

*

Converting to stoke

$$\begin{aligned} & 4.11 \times 10^{-7} \frac{\text{m}^2}{\text{s}} \times \frac{1 \text{ stoke}}{0.0001 \text{ m}^2/\text{s}} \\ & 4.11 \times 10^{-3} \text{ stoke} \\ & 4.11 \times 10^{-3} \text{ stoke} \end{aligned}$$

=

$$\begin{aligned} \nu &= \frac{\mu}{\rho} = \frac{\text{Pa} \cdot \text{s}}{\text{kg}/\text{m}^3} \\ &= \frac{\text{N}/\text{m}^2 \cdot \text{s}}{\text{kg}/\text{m}^3} \\ &= \text{m}^2/\text{s} \end{aligned}$$

* 7. Two large parallel surfaces 25mm apart and the space between them is filled with a liquid of viscosity $\mu = 0.958 \text{ Pa}\cdot\text{s}$. A thin plate of area $A = 0.37 \text{ m}^2$ is pulled at a constant speed $v = 0.3 \text{ m/s}$ if the plate is 8.4mm from one of the surfaces? What force is required to pull the plate at a constant speed of $v = 0.3 \text{ m/s}$ if the plate is 8.4mm from one of the surfaces?

Given: $\mu = 0.958 \text{ Pa}\cdot\text{s}$

Given: $\rho = 0.958$

$$\sum F_x = 0$$

$$F = F_1 + F_2$$

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$$\mu = \frac{T}{dV/dy}; \mu = \frac{F/A}{V/y}; \mu = \frac{FY}{AV}$$

$$* F = \frac{\mu AV}{Y}$$

$$F_1 = \frac{\mu AV}{Y_1} = \frac{0.958 (0.37)(0.3)}{16.6 \times 10^{-3}} = 6.41 \text{ N}$$

$$F_2 = \frac{\mu AV}{Y_2} = 12.66 \text{ N}$$

$$F_2 = \frac{\mu AV}{Y_2} = \frac{0.958 (0.37)(0.3)}{8.4 \times 10^{-3}} = 12.66 \text{ N}$$

$$F = 19.07 \text{ N}$$

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* An 18 kg slab slides down a 15° inclined plane on a film of oil with viscosity of 0.0814 Pa·s. If the contact area is 0.3 m², find the terminal velocity of the slab. Neglect air resistance.

$$\sum F_x = 0 \quad V = \frac{FY}{\mu A} = \frac{45.1(3 \times 10^{-3})}{0.0814(0.3)}$$

$$W_x - F = 0 \quad V = 5.61 \text{ m/s}$$

$$F = W_x \sin 15^\circ$$

$$F = 177.658 \sin 15^\circ = 45.7 \text{ N}$$

* 9. What is the value of the surface tension of a small water droplet of 0.3 mm in diameter, which is in contact with air if the pressure inside the droplet is 561 Pascal?

$$P = \frac{4\sigma}{d}$$

$$\sigma = \frac{Pd}{4} = \frac{561(0.3)(10^{-3})}{4}$$

$$\sigma = 0.042 \text{ N/m}$$